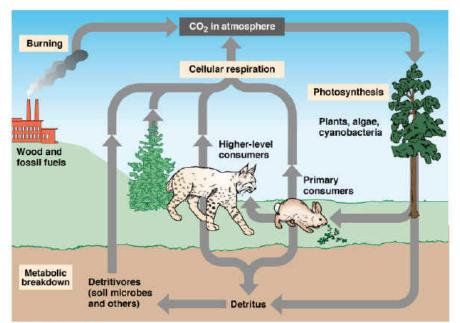
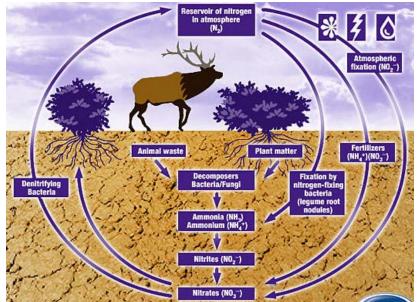
CARBON CYCLE



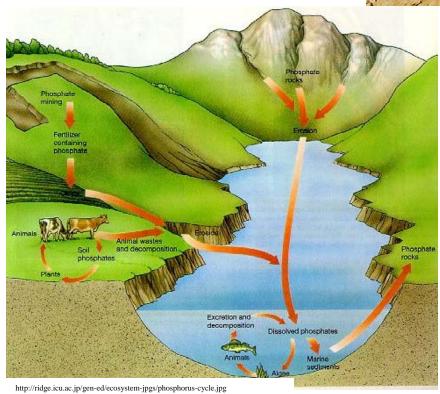
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NITROGEN CYCLE



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PHOSPHORUS CYCLE



Nutrient Cycles

Notes:

Carbon Cycle

* Carbon EXISTS in abiotic environment as:

- 1. Carbon dioxide $[CO_{2\,(gas)}]$ in the atmosphere
 - → dissolves in H₂O to form HCO₃
- 2. Carbonate rocks (limestone & $coral = CaCO_3$)
- 3. Deposits of coal, petroleum, and natural gas

 → derived from once living things
- 4. Dead organic matter (humus in the soil)

* Carbon ENTERS biotic environment through:

1. Photosynthesis: changes light energy to chemical energy

* Carbon RETURNS to atmosphere by:

- 1. Respiration \rightarrow CO₂
- 2. Decomposition / Decay
- 3. Burning

* Carbon Cycle and Humans:

- 1. Removal of photosynthesizing plants
- 2. Combustion of fossil fuels

Nitrogen Cycle

- * ~79% of air is N_2 gas
- * N is essential to plants and animals
- * Plants and animals can't use N_2 gas
- * Usable N: ammonia (NH₃) or nitrate (NO₃)

* Conversion of atmospheric N₂ to NH₃ and NO₃:

- → Nitrogen fixation
 - 1. Aquatic ecosystems: blue-green algae
 - 2. Terrestrial ecosystems: bacteria on root nodules of legumes (peas, beans, alfalfa, clover)
 - 3. Lightening

* Nitrogen RETURNS to soil by:

- 1. decomposition of once living things
 - → ammonifying bacteria + fungi
- 2. exists in soil as nitrate (NO₃-), nitrite (NO₂-), and ammonia (NH₃)

* Nitrogen returns to atmosphere by:

1. denitrifying bacteria

Nitrogen Cycle and Humans:

1. Nitrogen required for genetic materials (DNA, RNA, amino acids)

Phosphorus Cycle

- * Major environmental reservoir: rocks
- 1. Leaching: water dissolves phosphates in rocks and carries to lake, stream, etc.
- 2. Dissolved phosphate: used by plants and passed through food chain
- 3. Animals return phosphorus to environment by:
 - * excretion
 - * death and decay

Phosphorus Cycle and Humans:

- 1. Phosphates mined for fertilizers → returns P to soil
- 2. Erosion: P in soil and rocks washed away into water systems